

CONVENIENT SYNTHESSES OF 6-ARYLMETHYL- AND 6-(1-E-PROPENYL)-3-PYRIDINOLS

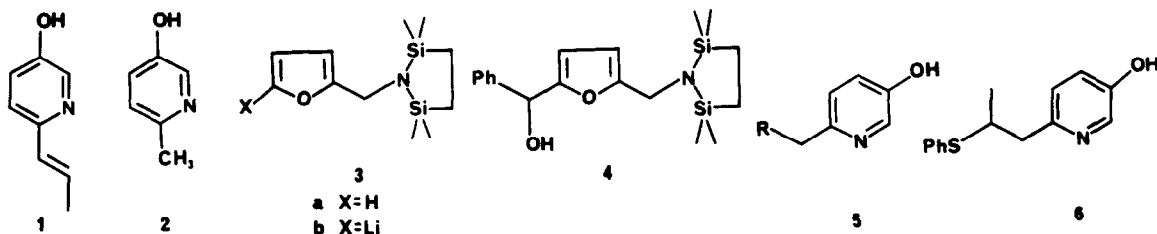
Anthony G.M. Barrett* and Suzanne A. Lebold

Department of Chemistry, Northwestern University, Evanston, Illinois 60208

ABSTRACT: 1-(2-Furylmethyl)-2,2,5,5-tetramethyl-1-aza-2,5-disilacyclopentane **3a** was reacted sequentially with *t*-butyllithium, an aldehyde and hydrochloric acid to produce the corresponding 6-substituted 3-pyridinol.

Recently we needed a convenient, simple, large scale synthesis of 6-substituted 3-pyridinol derivatives including 6-(1-E-propenyl)-3-pyridinol **1**¹. This molecule is important since it is an intermediate in the synthesis of the N-terminal amino acid of the nikkomycins, a group of potent antifungal agents. The related compound 6-methyl-3-pyridinol **2** is readily available from the acid mediated reaction of formaldehyde with 2-aminomethylfuran.² Additionally such ring expansion methodology can be used to prepare 2-alkyl and 2-aryl substituted 3-pyridinols from 2-furyl alkyl or aryl ketones and ammonia.³ In spite of all these studies the process is limited by the availability of the starting 2,5-disubstituted furans. Herein we report a simple procedure for converting 2-aminomethylfuran into the corresponding pyridine derivatives via C-5 lithiation.⁴

2-Aminomethylfuran was protected by reaction with 1,2-bis-(chlorodimethylsilyl)ethane⁵ and triethylamine in anhydrous dichloromethane solution to produce **3a**⁶ (83%). This material was cleanly metallated by reaction with *t*-butyllithium in THF and the resultant C-5 anion **3b** condensed with benzaldehyde. The adduct **4** was not isolated but was directly rearranged *in situ* by reaction with hydrochloric acid to produce 6-benzyl-3-pyridinol **5** (R=Ph)⁷ (37%).



The procedure was extended to the additional aromatic aldehydes **5** (R=4-MeC₆H₄; 42%), **5** (R=3-MeC₆H₄; 57%) and **5** (R=2-naphthyl; 36%). In an attempt to prepare **1**, propenal was reacted with **3b** but acidification gave only an intractable polymer. In contrast, **3b** reacted cleanly with 2-phenylthiopropional⁸ followed by hydrochloric acid to produce **6** together with the desired olefin **1**. Oxidation of the mixture using oxone⁹ followed by reflux in toluene solution cleanly gave **110** (14%). Although the yields of 6-substituted 3-pyridinols by this chemistry are only modest the procedure is both concise and experimentally straightforward to effect.

Experimental

1-(2-Furylmethyl)-2,2,5,5-tetramethyl-1-aza-2,5-disilacyclopentane. A solution of 1,2-bis-(chlorodimethylsilyl)ethane (6.36g) in anhydrous CH_2Cl_2 (11mL) was added to a stirred mixture of freshly distilled 2-aminomethylfuran (2.61mL) and dry triethylamine (10.28mL) in anhydrous CH_2Cl_2 (19mL) at 0°C under nitrogen. After the addition was complete, the mixture was warmed to room temperature and stirred for 2h. The triethylammonium chloride was filtered off, the residue evaporated and distilled to give 3a (5.87g, 83%): bp 48°C at 200mm Hg.

3-Hydroxy-6-(4-methylbenzyl)pyridine (general procedure). To a stirred solution of 3a (50.4mg) in anhydrous THF (5mL) at -78°C under nitrogen was slowly added *n*-butyllithium in hexane (1.7M; 1.37mL). The mixture was warmed to -42°C and stirring continued for 2h. 4-Methylbenzaldehyde (0.373mL) was added and the mixture stirred at room temperature for 1h. The solution was recooled to -78°C , quenched with hydrochloric acid (1M; 12.7mL) and refluxed (18h). The residue was cooled, diluted with ether and extracted with hydrochloric acid (3x25mL). The combined acidic extracts were washed with diethyl ether (50mL) and neutralized to pH7 with aqueous sodium hydroxide (1M). The solution was extracted with diethyl ether (3x50mL), the combined extracts were dried (MgSO_4) and evaporated to yield pure crystalline 3-hydroxy-6-(4-methylbenzyl)pyridine (0.175g, 42%).

ACKNOWLEDGEMENT: We thank the National Institutes of Health for support of this program (AI-22252), the Midwest Center for Mass Spectrometry, an NSF Regional Instrument Facility (CHE-8211164) for obtaining mass spectral data and Dr. Mark A. Russell for the preparation of a sample of 1 via 2 (reference 10).

References

1. W. Hass and W.A. König, Liebigs Ann. Chem., 1982, 1615.
2. N. Elming in Advances in Organic Chemistry, Methods and Results, Interscience Publishers Inc., New York, 1960, vol. 2, p. 67-115 and references cited therein.
3. For example see H. Leditschke, Chem. Ber., 1953, 86, 123. W. Gruber, Can. J. Chem., 1953, 31, 564.
4. For examples of furan α -lithiation see V. Ramanathan and R. Levine, J. Org. Chem., 1962, 27, 1216. D.W. Knight and A.P. Nott, J. Chem. Soc. Perkin Trans. 1, 1981, 1125.
5. For the use of 1,2-bis-(chlorodimethylsilyl)ethane in the protection of amino functionality see S. Djuric, J. Venit and P. Magnus, Tetrahedron Lett., 1981, 22, 1787.
6. All new compounds were fully characterized by spectroscopic methods, microanalyses and/or high resolution mass spectra.
7. Y. Kitaura, T. Oku, M. Hirai, T. Yamamoto, and M. Hashimoto, Eur. Pat. Appl. EP 157346 A2; Chem. Abstr., 1985, 104, 148759r.
8. R. Verhé, N. De Kimpe, L. De Buyck, and N. Schamp, Synthesis, 1984, 46.
9. B.M. Trost and D.P. Curran, Tetrahedron Lett., 1981, 1287.
10. The product 1 was identical with material that was prepared from 2 via sequential reaction with *n*-butyllithium, acetaldehyde, toluene-4-sulfonyl chloride, DBU, and potassium hydroxide.

(Received in USA 14 August 1987)